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Investigation of Langmuir Films of Fullerene Derivative Resulting by Addition of C_{60} to Tetracyanoethylene Oxide

V.R. NOVAK, S.L. VOROB'EVA, and I.V. MYAGKOV

Research Institute of Physical Problems, Zelenograd, Moscow, 103460, Russia

New fullerene-tetracyano-ethylene oxide derivative (C₆₀-TCNEO) was synthesized. The Langmuir layers of C₆₀-TCNEO on the water surface were investigated. The Langmuir-Blodgett films of C₆₀-TCNEO were prepared. The absorption spectra of obtained LB-films in UV and visible range (200 nm to 700 nm) and the dielectric constant at 1 kHz, 10 kHz, and 100 kHz were measured.

At present, the Langmuir-Blodgett (LB) films of fullerene C₆₀ and its derivatives is a subject of extensive studies, ¹⁻⁵ since fullerene is a molecular building block, which can be used to prepare new materials with interesting electronic properties.

In this paper we describe the synthesis of fullerene-tetracyano-ethylene oxide derivative (C₆₀-TCNEO) and the results of a study of its Langmuir monolayers on water surface and LB-films on solid substrates.

Synthesis of starting fullerene was carried out in our laboratory. An apparatus for production of carbon soot is analogous to that described in the literature but modified for the purpose of high efficiency by construction of a rotating graphite rod holder. The method described in was used for purification and separation.

NMR spectra were recorded on Varian XL-400 spectrometer at 400 MHz with tetramethylsilane as internal standard. IR spectra were determined in KBr by PYE UNICAM PU 9512 IR spectrometer. Mass-spectrum of compounds were recorded on Finnigan MAT 8430 mass-spectrometer. Samples were ionized by a xenon beam with energy of 8 kV from a matrix of m-nitrobenzyl alcohol.

Treatment of C₆₀ with tetracyanoethylene oxide in dichloroethane under 2h reflux results in the formation of a new compound 2 (C₆₀-TCNEO) (Scheme). Evaporation of solvent and separation by column chromatography (silica gel, mixture of hexane/toluene) result in compound 2 with 90 % yield. 50-fold excess of tetracyanoethylene was used. New compound is characterized by their ¹³C NMR, IR-and UV-spectra.

$$C_{60} + (NC)_2C --- C(CN)_2 ---- (NC)_2C --C_{60} --C(CN)_2$$

Scheme

The dependence of surface pressure versus area per molecule during compression of monolayer on water surface was investigated using a computer-controlled LB-trough "MDT-LB5" which had two compressing barriers. The surface pressure measurements were carried out with accuracy of $\pm 0.01 \, \text{mN/m}$.

LB-films were deposited onto substrates by the Langmuir-Schaefer method. The substrates had dimensions 12×24×1 mm³.

For study of dielectric properties we fabricated some samples in the form of thin-film capacitors with LB-films of C₆₀-TCNEO as dielectric layer. There were eight capacitors on a substrate. Each capacitor had area 1 mm². The sample capacitance was measured with an automatic capacitance bridge HP4270A.

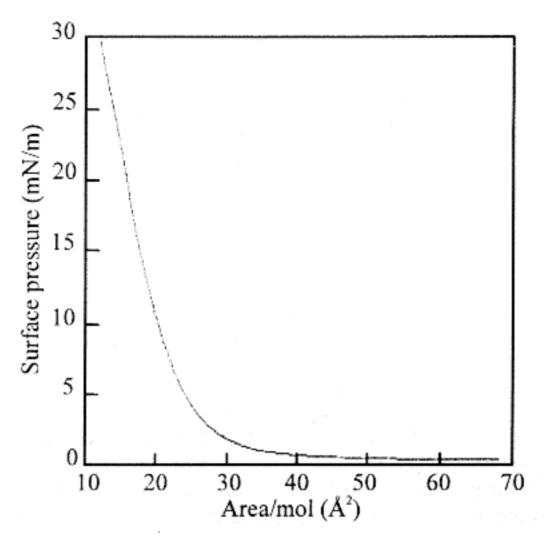


Figure 1 Dependence of surface pressure on area per molecule for the Langmuir layer of C₆₀-TCNEO on water surface.

The thickness of LB-films was measured with interference microscope "INTERFACE CARLZEISS".

Absorption spectra were recorded with LOMO CSVU-23 spectrophotometer.

A typical dependence of surface pressure versus area per molecule for C_{60} -TCNEO (π -A isotherm of compression) is presented in Figure 1 (dose of solution - 60 μ l, concentration of C_{60} -TCNEO 0.56 mg/ml).

The limiting area per molecule of C_{60} -TCNEO in layer obtained from linear approximation of the part of π -A curve at large values of surface pressure gives the value close to 25 Ų. This value, as in case of pure C_{60} , ^{1.5} is much smaller than that predicted from the X-ray diffraction studies of C_{60} (about 87 Ų). At indicates the formation of aggregates in C_{60} -TCNEO layer as in the case of pure C_{60} . It may be noted that π -A curves of C_{60} -TCNEO differ from the linear dependences of typical "condensed films", and its form at a low surface pressure (Figure 2) reminds more the corresponding dependences for "expanded films". In Figure 3 the surface pressure-area data are presented as surface pressure - molecular area product versus surface pressure. It may be seen that at low pressure $(\pi \approx 0.1 \div 1 \text{ mN/m})$ this dependence is described by linear law:

$$\pi(A-A_o) = qkT \tag{1}$$

which corresponds to the equation of state for two-dimensional gas.⁸⁻¹¹ In Equation 1 k is the Boltzmann constant, T is the absolute temperature, A_o is the effective area of one molecule and q is the parameter which characterizes the intensity of cohesive forces. Equation 1 may be given as:

$$\pi[(A/q)-(A_o/q)] = kT$$

were A/q is the area per one kinetic unit in a monolayer, A_o/q is the effective area of one kinetic unit and 1/q is the number of molecules in one kinetic unit. In our case the effective area per molecule in aggregate (A_o) is equal to 30 Å². This value is approximately three times smaller than the area of one molecule for C_{60} (87 Å²). Thus, the thickness of aggregates is approximately equal to three molecules of C_{60} -TCNEO. The average number of molecules in aggregate (1/q) is equal to 93. The average area of one aggregate is approximately 2790 Å² and the linear size is about 53 Å.

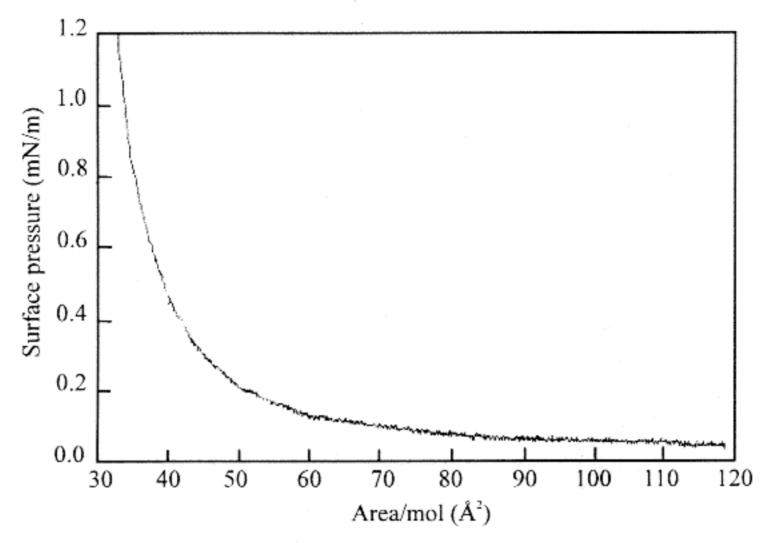


Figure 2 Dependence of surface pressure on area per molecule for the Langmuir layer of C₆₀-TCNEO on water surface at low surface pressure.

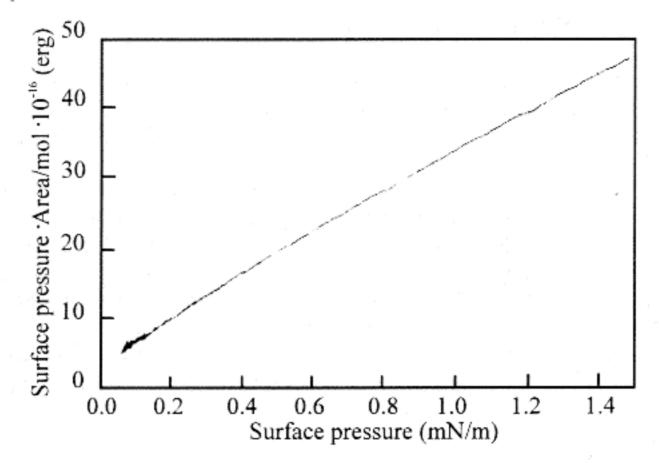


Figure 3 Surface pressure - molecular area product versus surface pressure for the Langmuir layer of C₆₀-TCNEO on water surface at low surface pressure.

The Langmuir layers were deposited onto a substrate from water surface at area per molecule $A = A_o$, where $A_o = 30 \text{ Å}^2$ is an affective area per molecule in an aggregate obtained from the surface pressure-area isotherm.

The electronic absorption spectra of LB-films of C_{60} -TCNEO on fused quartz were measured in UV and visible range (200 nm to 700 nm). The investigated films have absorption bands at $\lambda_1 = 215$ nm, $\lambda_2 = 265$ nm, and $\lambda_3 = 335$ nm.

The values of the dielectric constant of C_{60} -TCNEO are 3.29 \pm 0.05 at 1 kHz, 3.08 \pm 0.05 at 10 kHz and 2.96 \pm 0.05 at 100 kHz.

The effective thickness of one layer deposited on a substrate was determined by dividing the LB-film thickness into number of deposited layers. The resulting value for C_{60} -TCNEO is 27.7 Å \pm 0.2 Å. Thus, the effective thickness of one layer of C_{60} -TCNEO equals approximately three molecules. This result is in agreement with the data obtained from the surface pressure-area experiment.

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